# $C_{22}H_{36}O_2$

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# 3-Methoxy-16,16-dimethyl-5 $\alpha$ -androstan-17-one

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### Abstract

The structure of the title compound,  $C_{22}H_{36}O_2$ , is reported.

## Comment

The synthesis of the title compound, (I), has been reported in an earlier paper (Wölfling, Schneider & Dombi, 1988).



The A and B, B and C, and C and D rings are trans fused. Rings A, B and C adopt chair conformations, whereas ring D shows a conformation midway between a  $14\alpha$ envelope and a  $13\beta$ ,  $14\alpha$ -half chair. For structures of other  $5\alpha$ -androstane derivatives see Duax & Norton (1975) and Griffin, Duax & Weeks (1984).



Fig. 1. The atomic numbering scheme for the title compound.

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Crystal data C22H36O2  $M_r = 332.51$ Orthorhombic  $P2_{1}2_{1}2_{1}$ a = 6.5620 (10) Åb = 13.583 (3) Åc = 22.488 (4) Å V = 2004.4 (6) Å<sup>3</sup> Z = 4 $D_{\rm r} = 1.102 {\rm Mg m}^{-3}$ 

Experimental

Data collection Stoe AED-2 four-circle diffractometer Profile fitting collection method Absorption correction: none 2055 measured reflections 1769 independent reflections 1574 observed reflections  $[I > 2\sigma(I)]$ 

#### Refinement

C1

C2

C3

**O**3 C3

Refinement on  $F^2$ R(F) = 0.0363 $wR(F^2) = 0.0960$ S = 0.9681768 reflections 218 parameters H-atom parameters not refined Calculated weights  $w = 1/[\sigma^2(F_o^2) + (0.0610P)^2]$ + 0.5000P] where  $P = (F_o^2 + 2F_c^2)/3$ 6.1.1.4)  $(\Delta/\sigma)_{\rm max} = 0.009$ 

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters  $(Å^2)$ 

 $U_{\text{eq}} = (1/3) \sum_i \sum_j U_{ij} a_i^* a_i^* \mathbf{a}_i \cdot \mathbf{a}_j.$ 

x	у	z	$U_{eq}$
-0.1710 (6)	-0.5081 (2)	-0.17046 (13)	0.0542 (8)
-0.1458 (6)	-0.6103 (2)	-0.14299 (14)	0.0601 (9)
0.0256 (6)	-0.6124 (2)	-0.09872 (14)	0.0588 (9)
0.0338 (5)	-0.7096 (2)	-0.07388 (11)	0.0802 (8)
0.2192 (9)	-0.7317 (3)	-0.0455 (2)	0.111 (2)
-0.0051 (7)	-0.5341 (2)	-0.05120 (13)	0.0596 (9)
-0.0294 (5)	-0.4321 (2)	-0.07904 (12)	0.0485 (8)
-0.0393 (7)	-0.3510(2)	-0.03240 (13)	0.0610 (10)
-0.0396 (6)	-0.2493 (2)	0.06111 (12)	0.0551 (9)
-0.2041 (5)	-0.2385 (2)	-0.10876 (12)	0.0454 (8)
-0.1922 (5)	-0.3235 (2)	-0.15461 (12)	0.0433 (7)
-0.2058 (5)	-0.4262 (2)	-0.12412 (12)	0.0434 (7)
-0.3390 (6)	-0.3094 (2)	-0.20691 (12)	0.0546 (9)
-0.3254 (6)	-0.2081 (2)	-0.23612 (13)	0.0566 (9)
-0.3512 (5)	-0.1277 (2)	-0.18950 (13)	0.0480 (8)
-0.1893 (5)	-0.1407 (2)	-0.14117 (11)	0.0445 (7)
0.1946 (6)	-0.0435 (2)	-0.10730 (13)	0.0556 (9)
-0.2234 (5)	0.0359 (2)	-0.15601 (13)	0.0525 (8)
-0.3700 (7)	0.1181 (2)	-0.1383 (2)	0.0787 (11)
-0.0203 (6)	0.0820 (2)	-0.1742 (2)	0.0668 (10)
	$\begin{array}{c} x\\ -0.1710 (6)\\ -0.1458 (6)\\ 0.0256 (6)\\ 0.0338 (5)\\ 0.2192 (9)\\ -0.0051 (7)\\ -0.0294 (5)\\ -0.0393 (7)\\ -0.0294 (5)\\ -0.0396 (6)\\ -0.2041 (5)\\ -0.1922 (5)\\ -0.2058 (5)\\ -0.3390 (6)\\ -0.3254 (6)\\ -0.3254 (6)\\ -0.3254 (5)\\ -0.1893 (5)\\ -0.1893 (5)\\ -0.1946 (6)\\ -0.2234 (5)\\ -0.3700 (7)\\ -0.0203 (6) \end{array}$	$\begin{array}{cccc} x & y \\ -0.1710 (6) & -0.5081 (2) \\ -0.1458 (6) & -0.6103 (2) \\ 0.0256 (6) & -0.6124 (2) \\ 0.0338 (5) & -0.7096 (2) \\ 0.2192 (9) & -0.7317 (3) \\ -0.0051 (7) & -0.5341 (2) \\ -0.0294 (5) & -0.4321 (2) \\ -0.0393 (7) & -0.3510 (2) \\ -0.0396 (6) & -0.2493 (2) \\ -0.2041 (5) & -0.2385 (2) \\ -0.2041 (5) & -0.2385 (2) \\ -0.2058 (5) & -0.4262 (2) \\ -0.3390 (6) & -0.3094 (2) \\ -0.3512 (5) & -0.1277 (2) \\ -0.1893 (5) & -0.1277 (2) \\ -0.1893 (5) & -0.1407 (2) \\ -0.2234 (5) & 0.0359 (2) \\ -0.3700 (7) & 0.1181 (2) \\ -0.0203 (6) & 0.0820 (2) \\ \end{array}$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

Mo  $K\alpha$  radiation  $\lambda = 0.71072 \text{ Å}$ Cell parameters from 48 reflections  $\theta = 10 - 12.5^{\circ}$  $\mu = 0.068 \text{ mm}^{-1}$ T = 293 (2) KNeedles  $0.50 \times 0.40 \times 0.40$  mm Colorless

 $R_{\rm int} = 0.0108$  $\theta_{\text{max}} = 22.50^{\circ}$  $h = -1 \rightarrow 7$  $k = -14 \rightarrow 14$  $l = -24 \rightarrow 24$ 3 standard reflections frequency: 90 min intensity variation: none

 $\Delta \rho_{\rm max}$  = 0.119 e Å<sup>-3</sup>  $\Delta \rho_{\rm min} = -0.116 \ {\rm e} \ {\rm \AA}^{-3}$ Extinction correction:  $F_c^* = kF_c[1 + (0.001\chi$  $\times F_c^2 \lambda^3 / \sin 2\theta)]^{-1/4}$ Extinction coefficient:  $\chi = 0.012$  (2) Atomic scattering factors from International Tables for Crystallography (1992, Vol. C, Tables 4.2.6.8 and

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C17	-0.3084 (5)	-0.0230	(2)	-0.20887 (14)	0.0524 (8)		
017	-0.3305 (4)	0.0101	(2)	-0.25857 (10)	0.0719 (7)		
C18	-0.5715 (5)	-0.1271	(3)	-0.1660 (2)	0.0679 (10		
C19	-0.4161 (5)	-0.4413	(2)	-0.09522 (15)	0.0605 (9)		
Table 2. Selected geometric parameters (Å, °)							
C1–C2	2	1.528 (4)	C9-	-C10	1.557 (4)		
C1C1	10	1.542 (4)	C10	)—C19	1.539 (5)		
C2-C3	3	1.502 (5)	C11	I-C12	1.527 (4)		
C303	3	1.435 (4)	C12	2—C13	1.524 (4)		
C3-C4	1	1.521 (4)	C12	3—C17	1.513 (4)		
03-C	3'	1.406 (5)	C13	3—C14	1.530 (4)		
C4-C5	5	1.529 (4)	C13	3—C18	1.539 (5)		
C5-C6	5	1.522 (4)	C14	4—C15	1.524 (4)		
C5-C1	10	1.541 (4)	C1:	5—C16	1.549 (4)		
C6C7	7	1.526 (4)	C10	5—C16A	1.527 (5)		
C7-C8	8	1.528 (4)	C10	5—C16B	1.529 (5)		
C8-C	14	1.518 (4)	C10	5—C17	1.538 (4)		
C8-C9	9	1.550 (3)	C17	7—017	1.213 (3)		
C9C	11	1.532 (4)					
C2-C	1—C10	113.5 (2)	Cl	-C10-C9	109.9 (2)		
C3-C2	2C1	111.5 (3)	C1	2—C11—C9	113.9 (2)		
O3-C	3—C2	107.7 (3)	C1	3-C12-C11	110.1 (2)		
03-C	3—C4	112.0 (2)	CI	7—C13—C12	117.1 (2)		
C2C	3—C4	110.7 (3)	CI	7—C13—C14	100.6 (2)		
C3'-C	D3C3	114.0 (3)	CI	2-C13-C14	109.2 (2)		
C3-C4	4—C5	111.1 (2)	Cl	7—C13—C18	105.6 (3)		
C6-C	5—C4	112.2 (2)	C1	2-C13-C18	110.1 (3)		
C6-C	5—C10	112.6 (3)	CI	4C13C18	114.1 (2)		
C4C	5-C10	113.3 (3)	C8	-C14-C15	121.1 (2)		
C5C	6—C7	111.3 (2)	C8	-C14-C13	113.4 (2)		
C6C	7—C8	112.6 (3)	C1	5—C14—C13	103.8 (2)		
C140	C8—C7	112.1 (2)	Cl	4—C15—C16	104.6 (2)		
C14-0	C8—C9	109.2 (2)	C1	6A—C16—C16B	108.6 (3)		
C7C	8—C9	111.1 (2)	Cl	6A—C16—C17	110.7 (3)		
C11-0	C9—C8	112.7 (2)	CI	6BC16C17	108.8 (3)		
C11-0	C9—C10	114.5 (2)	CI	6A-C16-C15	113.7 (3)		
C8C	9—C10	111.8 (2)	Cl	6B—C16—C15	111.6 (3)		
C19—0	C10—C5	112.9 (2)	Cl	7-C16-C15	103.2 (2)		
C19-0	C10—C1	108.7 (3)	01	7-C17-C13	126.2 (3)		
C5-C	10—C1	107.2 (2)	01	7-C17-C16	124.2 (3)		
C19-0	C10—C9	110.9 (3)	CI	3-C17-C16	109.5 (2)		
C5-C	:10—C9	107.1 (2)					

Data collection: *DIF4* (Stoe & Cie, 1989*a*). Cell refinement: *DIF4*. Data reduction: *REDU4* (Stoe & Cie, 1989*b*). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1990). Program(s) used to refine structure: *SHELXL92* (Sheldrick, 1992).

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Lists of structure factors, anisotropic displacement parameters and H-atom coordinates have been deposited with the IUCr (Reference: KA1056). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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# An Enantiomerically Pure 4-Oxo- $\alpha$ -amino Acid Derivative

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#### Abstract

In the molecule of benzyl 2(S)-[(*tert*-butoxycarbonyl)amino]-4-oxo-6-phenyl-5(*E*)-hexenoate, C<sub>24</sub>H<sub>27</sub>NO<sub>5</sub>, the amine NH group does not form intramolecular hydrogen bonds with any of the three carbonyl groups present, even the most favourably positioned one, which would give a six-membered ring. Instead, weak intermolecular hydrogen-bonding interactions occur between the NH and CO groups of the amide units in adjacent molecules, linking them into a chain.

#### Comment

Enantiomerically pure  $\alpha$ -amino acids can be prepared in various ways (Williams, 1989). Recent research into methods which do not involve the formation of one of the four bonds to the  $\alpha$  centre and which instead retain the chirality already established at this centre, has been successful in producing a wide range of protected 4-oxo- $\alpha$ -amino acids and protected phenylalanine analogues from an organozinc intermediate (Jackson, Wishart, Wood, James & Wythes, 1992). The title compound, (I), is one of these 4-oxo- $\alpha$ -amino acid derivatives. Its structure has been determined in order to confirm the enantiomeric purity and to investigate the hydrogen bonding which may be expected to occur between the amino group and one of the three carbonyl groups within the molecule.



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